

N,N'-Bis(pyridin-3-yl)terephthalamide–terephthalic acid (1/1)

Ji-lin Lu,* Xue-wen Liu, Lin Li, Yuan-dao Chen and Guang-yu Shen

College of Chemistry and Chemical Engineering, Hunan University of Arts and Science, ChangDe, Hunan province 415000, People's Republic of China
Correspondence e-mail: lu_j_l@163.com

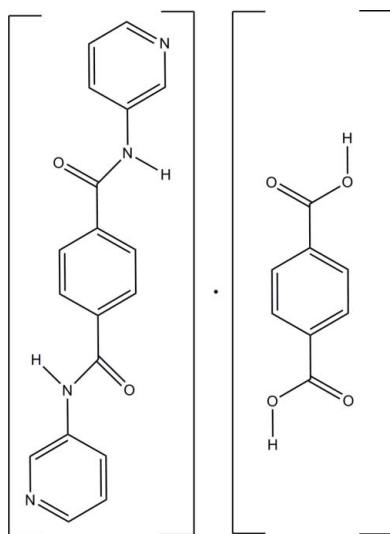
Received 17 September 2011; accepted 10 October 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.095; data-to-parameter ratio = 11.3.

In the title compound, $\text{C}_{18}\text{H}_{14}\text{N}_4\text{O}_2 \cdot \text{C}_8\text{H}_6\text{O}_4$, both types of molecule lie on inversion centers. In the *N,N'*-bis(pyridin-3-yl)terephthalamide molecule, the pyridine ring forms a dihedral angle of 11.33 (9)° with the central benzene ring. In the crystal, $\text{N}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{N}$ hydrogen bonds connect the components into a three-dimensional network.

Related literature

For related structures, see: Xiao *et al.* (2011), Wang *et al.* (2009).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{14}\text{N}_4\text{O}_2 \cdot \text{C}_8\text{H}_6\text{O}_4$
 $M_r = 484.46$
Monoclinic, $P2_1/c$
 $a = 11.0001$ (3) Å
 $b = 10.8080$ (2) Å
 $c = 9.6903$ (2) Å
 $\beta = 106.830$ (2)°

$V = 1102.73$ (4) Å³
 $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 0.11$ mm⁻¹
 $T = 296$ K
 $0.25 \times 0.24 \times 0.22$ mm

Data collection

Bruker SMART CCD diffractometer
Absorption correction: multi-scan (SABADS; Sheldrick, 1996)
 $T_{\min} = 0.974$, $T_{\max} = 0.977$

8117 measured reflections
1939 independent reflections
1640 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.095$
 $S = 1.07$
1939 reflections
171 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.14$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.19$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H \cdots A$ | $D-H$ | $H \cdots A$ | $D \cdots A$ | $D-H \cdots A$ |
|---|----------|--------------|--------------|----------------|
| $\text{N2}-\text{H2N} \cdots \text{O1}^{\text{i}}$ | 0.89 (2) | 1.98 (2) | 2.8616 (18) | 171.3 (18) |
| $\text{O2}-\text{H1N} \cdots \text{N1}^{\text{ii}}$ | 1.00 (3) | 1.69 (3) | 2.6938 (19) | 178 (2) |

Symmetry codes: (i) $x, -y + \frac{1}{2}, z + \frac{1}{2}$; (ii) $x, y + 1, z - 1$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINTE* (Bruker, 1998); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5338).

References

- Brandenburg, K. (1999). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
Bruker (1998). *SMART* and *SAINTE*. Bruker AXS Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Wang, S., Yang, T., Li, Z. & Yu, X. (2009). *Acta Cryst.* **E65**, o2198.
Xiao, W., Xue, R. & Yin, Y. (2011). *Acta Cryst.* **E67**, o1333.

supplementary materials

Acta Cryst. (2011). E67, o3064 [doi:10.1107/S1600536811041596]

N,N'-Bis(pyridin-3-yl)terephthalamide-terephthalic acid (1/1)

J. Lu, X. Liu, L. Li, Y. Chen and G. Shen

Comment

Pyridine amide derivatives and carboxylic acids easily form hydrogen bonds therefore they are useful to construct supramolecular structures (e.g. Xiao *et al.*, 2011; Wang *et al.*, 2009). Herein, we use *N,N'*-di(pyridin-3-yl)terephthalamide and terephthalic acid to construct a supramolecular compound. The crystal structure of the title compound is presented herein.

The molecular structure of the title compound is shown in Fig. 1. The symmetry unique pyridine ring forms a dihedral angle of 11.33 (9)° with the central benzene ring. In the crystal, N—H···O and O—H···N hydrogen bonds connect the components of the structure into a three dimensional network (Fig. 2).

Experimental

N,N'-di(pyridin-3-yl)terephthalamide (0.2 mmol) and terephthalic acid (0.2 mmol) was sealed in a teflon reactor with 6 mL water, and heated at 433 K for 2 days, and then cooled to room temperature. The single crystals were obtained by slow evaporation.

Refinement

H atoms bonded to C atoms were placed in calculated positions with C—H = 0.93 Å and included using a riding-model approximation with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. H atoms bonded to O and N atoms were refined independently with isotropic displacement parameters.

Figures

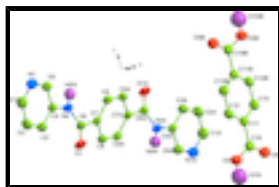


Fig. 1. The molecular structure of the title compound with displacement ellipsoids drawn at the 50% probability level (symmetry code; (A) $-x + 1, -y + 1, -z + 1$; (B) $-x + 2, -y + 2, -z + 1$). Hydrogen atoms bonded to C atoms are not shown.

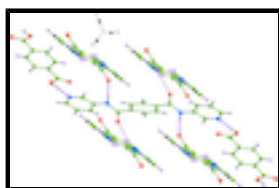


Fig. 2. Part of the crystal structure with hydrogen bonds shown as pink dashed lines. H atoms are purple.

N,N'-Bis(pyridin-3-yl)terephthalamide; terephthalic acid

Crystal data

| | |
|--------------------------------------|---|
| $C_{18}H_{14}N_4O_2 \cdot C_8H_6O_4$ | $F(000) = 504$ |
| $M_r = 484.46$ | $D_x = 1.459 \text{ Mg m}^{-3}$ |
| Monoclinic, $P2_1/c$ | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| Hall symbol: -P 2ybc | Cell parameters from 2279 reflections |
| $a = 11.0001 (3) \text{ \AA}$ | $\theta = 2.7\text{--}25.2^\circ$ |
| $b = 10.8080 (2) \text{ \AA}$ | $\mu = 0.11 \text{ mm}^{-1}$ |
| $c = 9.6903 (2) \text{ \AA}$ | $T = 296 \text{ K}$ |
| $\beta = 106.830 (2)^\circ$ | Block, colourless |
| $V = 1102.73 (4) \text{ \AA}^3$ | $0.25 \times 0.24 \times 0.22 \text{ mm}$ |
| $Z = 2$ | |

Data collection

| | |
|---|--|
| Bruker SMART CCD diffractometer | 1939 independent reflections |
| Radiation source: fine-focus sealed tube graphite | 1640 reflections with $I > 2\sigma(I)$ |
| φ and ω scans | $R_{\text{int}} = 0.035$ |
| Absorption correction: multi-scan (SABADS; Sheldrick, 1996) | $\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 1.9^\circ$ |
| $T_{\text{min}} = 0.974$, $T_{\text{max}} = 0.977$ | $h = -13 \rightarrow 12$ |
| 8117 measured reflections | $k = -12 \rightarrow 11$ |
| | $l = -11 \rightarrow 11$ |

Refinement

| | |
|---------------------------------|--|
| Refinement on F^2 | Primary atom site location: structure-invariant direct methods |
| Least-squares matrix: full | Secondary atom site location: difference Fourier map |
| $R[F^2 > 2\sigma(F^2)] = 0.040$ | Hydrogen site location: inferred from neighbouring sites |
| $wR(F^2) = 0.095$ | H atoms treated by a mixture of independent and constrained refinement |
| $S = 1.07$ | $w = 1/[\sigma^2(F_o^2) + (0.0378P)^2 + 0.3019P]$ |
| 1939 reflections | where $P = (F_o^2 + 2F_c^2)/3$ |
| 171 parameters | $(\Delta/\sigma)_{\text{max}} < 0.001$ |
| 0 restraints | $\Delta\rho_{\text{max}} = 0.14 \text{ e \AA}^{-3}$ |
| | $\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$ |

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations

between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|---------------|--------------|----------------------------------|
| C10 | 0.91350 (16) | 1.01651 (17) | 0.18960 (19) | 0.0385 (4) |
| C11 | 0.95943 (16) | 1.00643 (16) | 0.35009 (18) | 0.0358 (4) |
| C8 | 0.58897 (16) | 0.45790 (15) | 0.62372 (17) | 0.0336 (4) |
| H8 | 0.6489 | 0.4302 | 0.7068 | 0.040* |
| C7 | 0.54563 (16) | 0.37897 (14) | 0.50665 (16) | 0.0308 (4) |
| C5 | 0.74809 (16) | 0.04149 (16) | 0.79727 (18) | 0.0367 (4) |
| H5 | 0.7580 | 0.1028 | 0.8671 | 0.044* |
| C3 | 0.66694 (17) | -0.02113 (16) | 0.55212 (18) | 0.0381 (4) |
| H3 | 0.6219 | -0.0058 | 0.4565 | 0.046* |
| C6 | 0.59441 (16) | 0.25022 (15) | 0.50444 (17) | 0.0339 (4) |
| C9 | 0.45681 (17) | 0.42247 (15) | 0.38296 (17) | 0.0351 (4) |
| H9 | 0.4279 | 0.3704 | 0.3038 | 0.042* |
| C12 | 1.01793 (16) | 0.89971 (16) | 0.41856 (19) | 0.0388 (4) |
| H12 | 1.0304 | 0.8326 | 0.3642 | 0.047* |
| C4 | 0.68076 (15) | 0.06989 (15) | 0.65646 (17) | 0.0313 (4) |
| C13 | 0.94247 (17) | 1.10642 (16) | 0.43264 (19) | 0.0399 (4) |
| H13 | 0.9041 | 1.1782 | 0.3874 | 0.048* |
| C2 | 0.72163 (18) | -0.13483 (16) | 0.5939 (2) | 0.0429 (5) |
| H2 | 0.7154 | -0.1971 | 0.5259 | 0.051* |
| C1 | 0.78552 (18) | -0.15624 (17) | 0.7362 (2) | 0.0460 (5) |
| H1 | 0.8204 | -0.2341 | 0.7630 | 0.055* |
| N1 | 0.79926 (14) | -0.06916 (14) | 0.83764 (15) | 0.0423 (4) |
| N2 | 0.62725 (14) | 0.18931 (13) | 0.63090 (14) | 0.0338 (3) |
| H2N | 0.6221 (18) | 0.2299 (18) | 0.709 (2) | 0.048 (5)* |
| O1 | 0.60228 (13) | 0.20383 (11) | 0.39151 (12) | 0.0476 (4) |
| O2 | 0.91449 (14) | 0.91030 (12) | 0.12264 (15) | 0.0524 (4) |
| H1N | 0.874 (3) | 0.918 (3) | 0.016 (3) | 0.095 (9)* |
| O3 | 0.87627 (13) | 1.11288 (12) | 0.12828 (13) | 0.0501 (4) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|-------------|------------|------------|
| C10 | 0.0381 (9) | 0.0349 (10) | 0.0398 (10) | -0.0038 (8) | 0.0069 (8) | 0.0027 (8) |
| C11 | 0.0364 (9) | 0.0332 (10) | 0.0353 (9) | -0.0037 (7) | 0.0063 (7) | 0.0034 (7) |
| C8 | 0.0435 (9) | 0.0310 (9) | 0.0255 (8) | 0.0035 (7) | 0.0086 (7) | 0.0045 (7) |
| C7 | 0.0437 (9) | 0.0256 (9) | 0.0255 (8) | 0.0017 (7) | 0.0136 (7) | 0.0033 (7) |
| C5 | 0.0458 (10) | 0.0313 (10) | 0.0321 (9) | 0.0041 (8) | 0.0100 (8) | 0.0008 (7) |

supplementary materials

| | | | | | | |
|-----|-------------|-------------|-------------|-------------|------------|-------------|
| C3 | 0.0485 (10) | 0.0328 (10) | 0.0317 (9) | 0.0024 (8) | 0.0094 (8) | -0.0004 (7) |
| C6 | 0.0472 (10) | 0.0294 (9) | 0.0273 (9) | 0.0014 (7) | 0.0143 (7) | 0.0014 (7) |
| C9 | 0.0514 (10) | 0.0279 (9) | 0.0258 (8) | 0.0002 (7) | 0.0109 (8) | -0.0011 (7) |
| C12 | 0.0434 (10) | 0.0313 (9) | 0.0388 (10) | -0.0002 (8) | 0.0072 (8) | -0.0022 (8) |
| C4 | 0.0384 (9) | 0.0259 (9) | 0.0311 (9) | 0.0028 (7) | 0.0123 (7) | 0.0026 (7) |
| C13 | 0.0430 (10) | 0.0312 (10) | 0.0408 (10) | 0.0023 (8) | 0.0047 (8) | 0.0036 (8) |
| C2 | 0.0539 (11) | 0.0282 (10) | 0.0452 (11) | 0.0045 (8) | 0.0122 (9) | -0.0059 (8) |
| C1 | 0.0530 (11) | 0.0311 (10) | 0.0509 (12) | 0.0092 (8) | 0.0105 (9) | 0.0034 (9) |
| N1 | 0.0493 (9) | 0.0362 (9) | 0.0377 (8) | 0.0086 (7) | 0.0066 (7) | 0.0049 (7) |
| N2 | 0.0533 (9) | 0.0253 (8) | 0.0241 (7) | 0.0066 (6) | 0.0132 (6) | 0.0012 (6) |
| O1 | 0.0845 (10) | 0.0333 (7) | 0.0306 (7) | 0.0124 (6) | 0.0255 (6) | 0.0033 (5) |
| O2 | 0.0773 (10) | 0.0365 (8) | 0.0357 (8) | 0.0039 (7) | 0.0044 (7) | 0.0003 (6) |
| O3 | 0.0656 (9) | 0.0373 (8) | 0.0401 (7) | 0.0025 (6) | 0.0037 (6) | 0.0063 (6) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|------------------------|-------------|----------------------------|-------------|
| C10—O3 | 1.210 (2) | C6—O1 | 1.2293 (19) |
| C10—O2 | 1.320 (2) | C6—N2 | 1.345 (2) |
| C10—C11 | 1.494 (2) | C9—C8 ⁱ | 1.382 (2) |
| C11—C13 | 1.389 (2) | C9—H9 | 0.9300 |
| C11—C12 | 1.392 (2) | C12—C13 ⁱⁱ | 1.382 (2) |
| C8—C9 ⁱ | 1.382 (2) | C12—H12 | 0.9300 |
| C8—C7 | 1.389 (2) | C4—N2 | 1.410 (2) |
| C8—H8 | 0.9300 | C13—C12 ⁱⁱ | 1.382 (2) |
| C7—C9 | 1.391 (2) | C13—H13 | 0.9300 |
| C7—C6 | 1.494 (2) | C2—C1 | 1.374 (3) |
| C5—N1 | 1.331 (2) | C2—H2 | 0.9300 |
| C5—C4 | 1.385 (2) | C1—N1 | 1.337 (2) |
| C5—H5 | 0.9300 | C1—H1 | 0.9300 |
| C3—C2 | 1.377 (2) | N2—H2N | 0.89 (2) |
| C3—C4 | 1.387 (2) | O2—H1N | 1.00 (3) |
| C3—H3 | 0.9300 | | |
| O3—C10—O2 | 123.83 (16) | C8 ⁱ —C9—H9 | 119.6 |
| O3—C10—C11 | 122.57 (16) | C7—C9—H9 | 119.6 |
| O2—C10—C11 | 113.58 (15) | C13 ⁱⁱ —C12—C11 | 119.97 (16) |
| C13—C11—C12 | 119.40 (16) | C13 ⁱⁱ —C12—H12 | 120.0 |
| C13—C11—C10 | 118.80 (15) | C11—C12—H12 | 120.0 |
| C12—C11—C10 | 121.80 (16) | C5—C4—C3 | 118.38 (15) |
| C9 ⁱ —C8—C7 | 120.14 (15) | C5—C4—N2 | 116.91 (15) |
| C9 ⁱ —C8—H8 | 119.9 | C3—C4—N2 | 124.68 (15) |
| C7—C8—H8 | 119.9 | C12 ⁱⁱ —C13—C11 | 120.63 (16) |
| C8—C7—C9 | 119.05 (15) | C12 ⁱⁱ —C13—H13 | 119.7 |
| C8—C7—C6 | 122.97 (14) | C11—C13—H13 | 119.7 |
| C9—C7—C6 | 117.92 (14) | C1—C2—C3 | 119.90 (17) |
| N1—C5—C4 | 123.26 (16) | C1—C2—H2 | 120.1 |
| N1—C5—H5 | 118.4 | C3—C2—H2 | 120.1 |
| C4—C5—H5 | 118.4 | N1—C1—C2 | 122.36 (17) |

| | | | |
|-------------------------------|--------------|-------------------------------|--------------|
| C2—C3—C4 | 118.19 (16) | N1—C1—H1 | 118.8 |
| C2—C3—H3 | 120.9 | C2—C1—H1 | 118.8 |
| C4—C3—H3 | 120.9 | C5—N1—C1 | 117.91 (15) |
| O1—C6—N2 | 122.81 (16) | C6—N2—C4 | 126.60 (14) |
| O1—C6—C7 | 120.72 (14) | C6—N2—H2N | 117.7 (12) |
| N2—C6—C7 | 116.47 (14) | C4—N2—H2N | 115.4 (12) |
| C8 ⁱ —C9—C7 | 120.81 (15) | C10—O2—H1N | 111.7 (16) |
| O3—C10—C11—C13 | 9.3 (3) | N1—C5—C4—C3 | -0.6 (3) |
| O2—C10—C11—C13 | -169.05 (16) | N1—C5—C4—N2 | 177.29 (16) |
| O3—C10—C11—C12 | -171.05 (17) | C2—C3—C4—C5 | -0.4 (2) |
| O2—C10—C11—C12 | 10.6 (2) | C2—C3—C4—N2 | -178.11 (16) |
| C9 ⁱ —C8—C7—C9 | 0.6 (3) | C12—C11—C13—C12 ⁱⁱ | -0.6 (3) |
| C9 ⁱ —C8—C7—C6 | 177.70 (15) | C10—C11—C13—C12 ⁱⁱ | 179.14 (16) |
| C8—C7—C6—O1 | -146.83 (17) | C4—C3—C2—C1 | 1.3 (3) |
| C9—C7—C6—O1 | 30.4 (2) | C3—C2—C1—N1 | -1.3 (3) |
| C8—C7—C6—N2 | 33.8 (2) | C4—C5—N1—C1 | 0.6 (3) |
| C9—C7—C6—N2 | -149.04 (16) | C2—C1—N1—C5 | 0.3 (3) |
| C8—C7—C9—C8 ⁱ | -0.6 (3) | O1—C6—N2—C4 | 3.8 (3) |
| C6—C7—C9—C8 ⁱ | -177.85 (15) | C7—C6—N2—C4 | -176.86 (15) |
| C13—C11—C12—C13 ⁱⁱ | 0.6 (3) | C5—C4—N2—C6 | 157.60 (17) |
| C10—C11—C12—C13 ⁱⁱ | -179.13 (16) | C3—C4—N2—C6 | -24.7 (3) |

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+2, -y+2, -z+1$.

Hydrogen-bond geometry (\AA , $^\circ$)

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|-----------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| N2—H2N \cdots O1 ⁱⁱⁱ | 0.89 (2) | 1.98 (2) | 2.8616 (18) | 171.3 (18) |
| O2—H1N \cdots N1 ^{iv} | 1.00 (3) | 1.69 (3) | 2.6938 (19) | 178 (2) |

Symmetry codes: (iii) $x, -y+1/2, z+1/2$; (iv) $x, y+1, z-1$.

Fig. 1

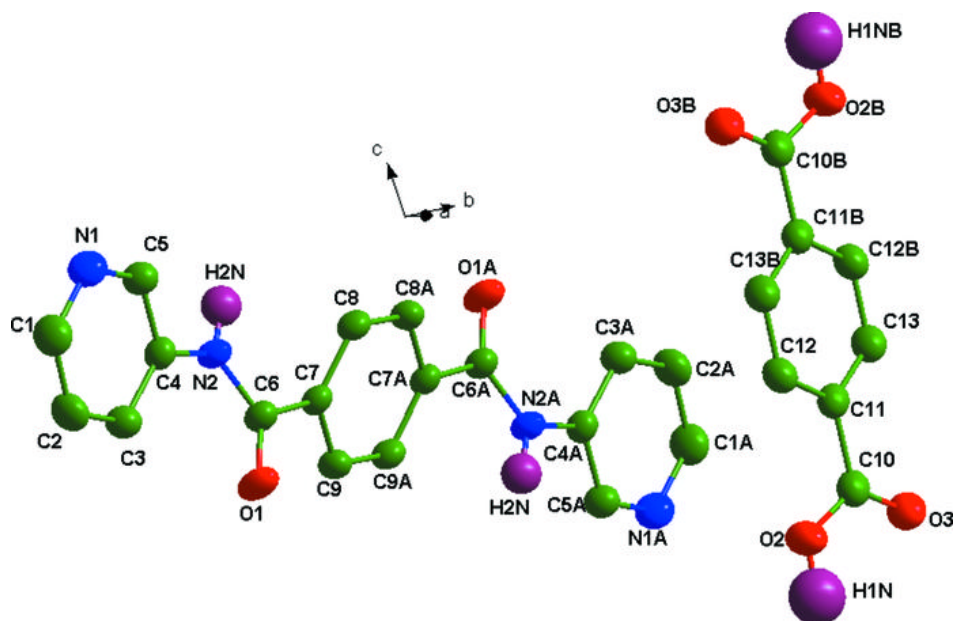


Fig. 2

